

**Critical raw materials, conflict minerals and
shortfall of elements used in electronic and
electrical devices. An approach taking into
account recycling and recovery of materials**

Topic: Robots and society, ethics, and
legal issues

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Glossary

Critical raw materials (CRMs): raw materials of a high importance to the economy of the EU and whose supply is associated with a high risk.

- **Economic Importance (EI)** and **Supply Risk (SR)** are used to determine the criticality of the material for the EU. The list of CRMs is established on the basis of the raw materials which reach or exceed the thresholds for both parameters.

Conflict minerals: due to internal conflict situations in a few mining regions in the world, there is a risk of armed conflicts being financed by the proceeds from the sale of minerals.

- **Columbite-tantalite** ('coltan'; commodity for niobium and tantalum extraction), incl. as a component of condensers; used for the production of steel (Niobium)
- **Cassiterite** (black tin; commodity for tin extraction), incl. as a component of catalytic converters, semiconductors, alloys; used in soldering processes, as a stabiliser for PVC, in packagings (food cans)
- **Wolframite** (commodity for tungsten extraction), incl. as a component of alloys in the automotive industry; used in microelectrodes, 'tungsten wire'
- **Gold**, incl. in jewellery making, the electronics, optical and medical industries, as coins and ingots in payment transactions

Glossary

Shortfall / endangered elements: Of the 118 elements that make up everything—from the compounds in a chemist’s arsenal to consumer products on the shelf—44 will face supply limitations in the coming years. These critical elements include rare earth elements, precious metals, and phosphorus.

- There are 9 elements (He, Zn, Ga, Ge, As, Ag, In, Te, Hf) for which there is concern that there is a serious threat to their supply within the next 100 years
- There are 7 elements (Ru, Rh, Ta, Os, Ir, Pt, U) for which there is a rising threat due to increased use.

Critical Raw Materials

- **Economic Importance (EI):** looks in detail at the allocation of raw materials to end-uses based on industrial applications.
- **Supply risk (SR):** looks at the country-level concentration of global production of primary raw materials and sourcing to the EU, the governance of supplier countries, including environmental aspects, the contribution of recycling (i.e. secondary raw materials), substitution, EU import reliance and trade restrictions in third countries.
- The 2020 EU list of CRMs contains 30 materials
 - 14 materials in 2011
 - 20 materials in 2014
 - 27 materials in 2017

Critical Raw Materials List

Antimony	Hafnium	Phosphorus
Baryte	Heavy Rare Earth Elements (HREE): Dy, Er, Eu, Gd, Ho, Lu, Tb, Tm, Yb, Y	Scandium
Beryllium	Light Rare Earth Elements (LREE): Ce, La, Nd, Pr, Sm	Silicon metal
Bismuth	Indium	Tantalum
Borate	Magnesium	Tungsten
Cobalt	Natural Graphite	Vanadium
Coking Coal	Natural Rubber	Bauxite
Fluorspar	Niobium	Lithium
Gallium	Platinum Group Metals (PGM): Ir, Pd, Pt, Rh, Ru	Titanium
Germanium	Phosphate rock	Strontium

CRMs – selected uses

CRM	Stage	Main global producers	Main EU sourcing countries	Import reliance	Selected uses
Be	E	United States (88%) China (8%) Madagascar (2%)	n/a	n/a	- Electronic and Communications Equipment - Automotive, aero-space and defense components
Borate	E	Turkey (42%) United States (24%) Chile (11%)	Turkey (98%)	100%	- High performance glass - Fertilisers - Permanent magnets
Co	E	Congo (59%) China (7%) Canada (5%)	Congo (68%) Finland (14%) French Guiana (5%)	86%	- Batteries - Super alloys - Catalysts - Magnets

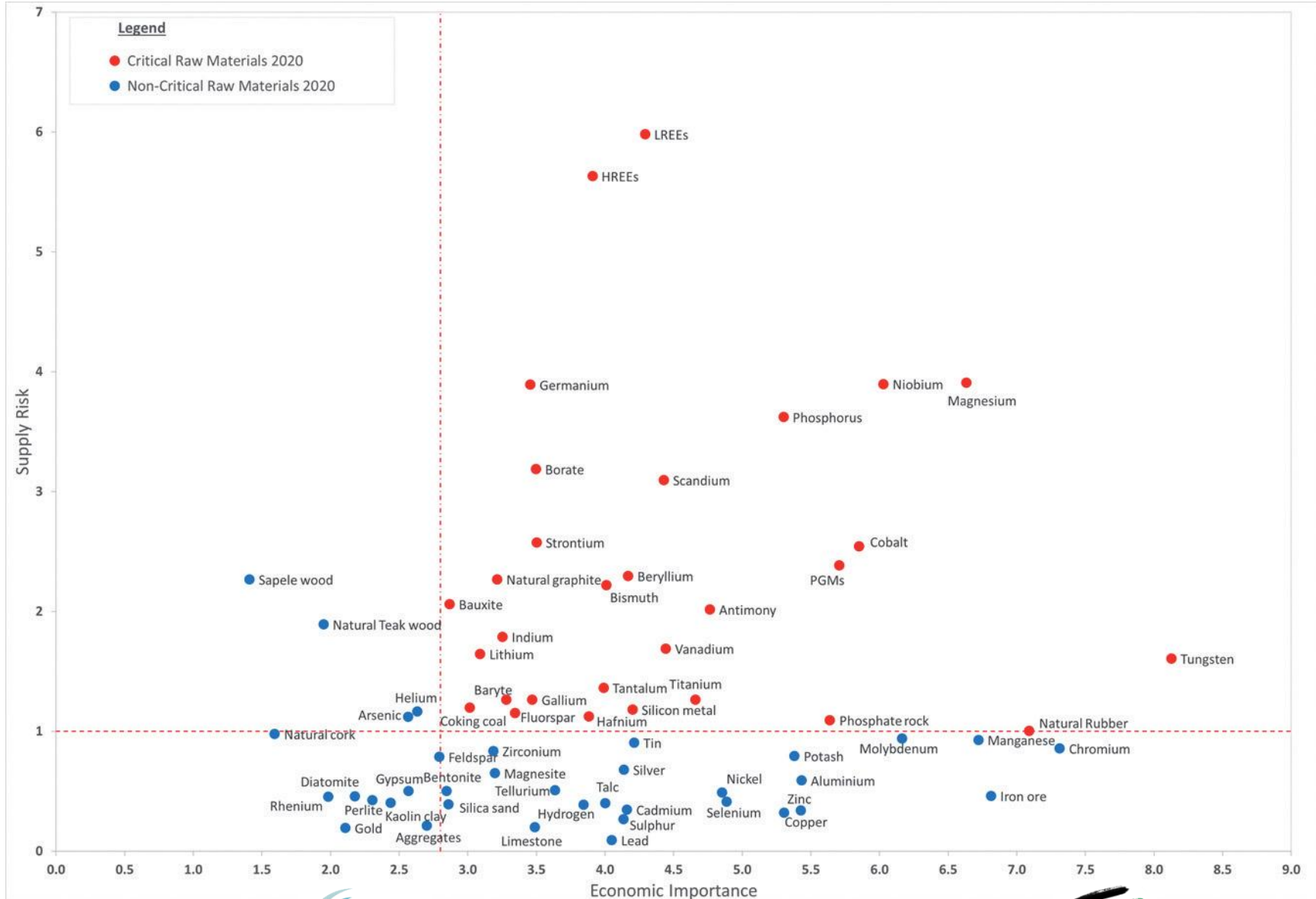
CRMs – selected uses

CRM	Stage	Main global producers	Main EU sourcing countries	Import reliance	Selected uses
Ga	P	China (80%) Germany (8%) Ukraine (5%)	Germany (35%) UK (28%) China (27%) Hungary (2%)	31%	- Semiconductors - Photovoltaic cells
Ge	P	China (80%) Finland (10%) Russia (5%)	Finland (51%) China (17%) UK (11%)	31%	- Optical fibers and Infrared optics - Satellite solar cells - Polymerisation catalysts
In	P	China (48%) Korea, Rep. (21%) Japan (8%)	France (28%) Belgium (23%) UK (12%) Germany (10%) Italy (5%)	0%	- Flat panel displays - Photovoltaic cells and photonics - Solders

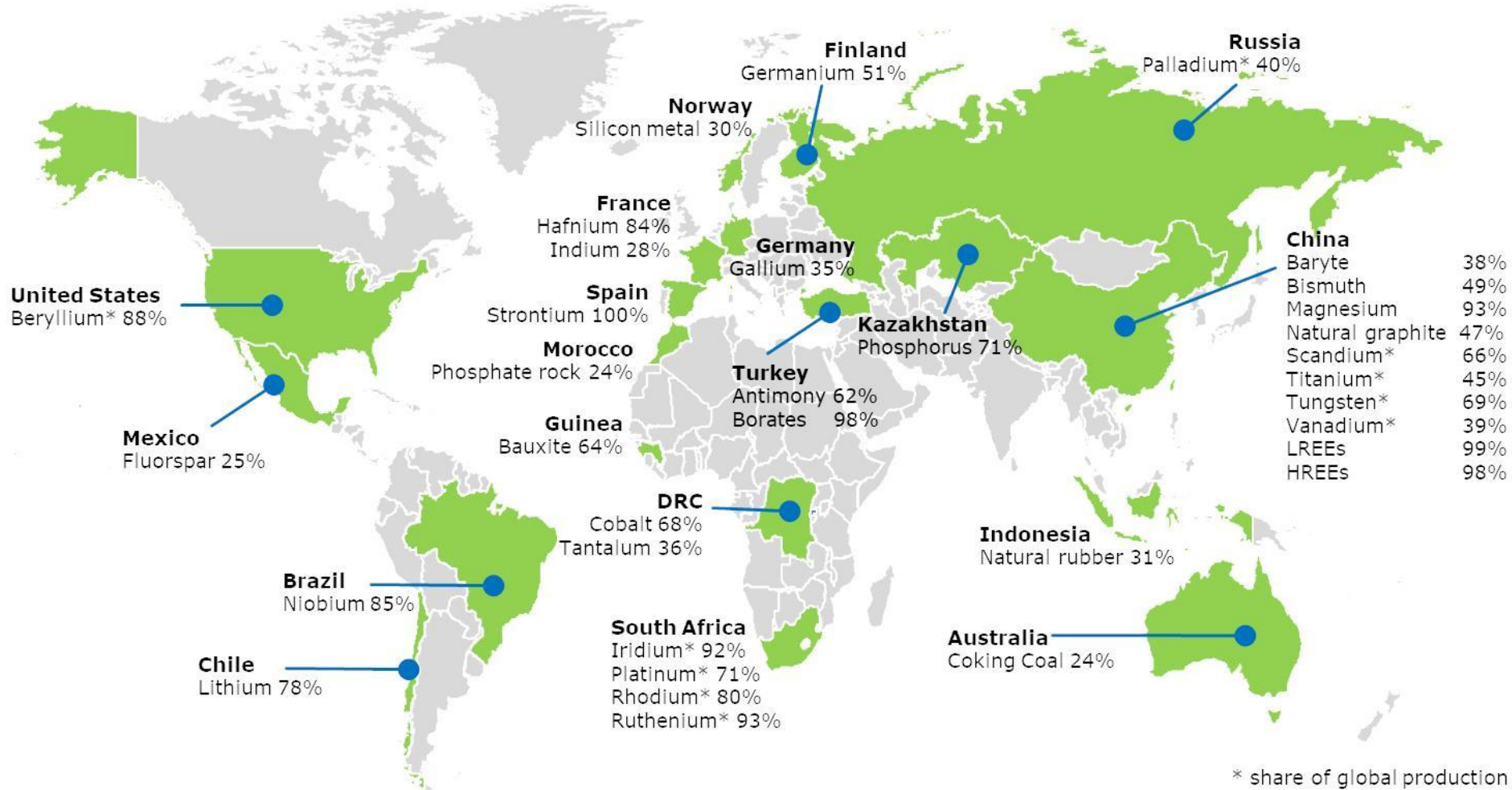
CRMs – selected uses

CRM	Stage	Main global producers	Main EU sourcing countries	Import reliance	Selected uses
PGMs	P	South Africa (84%) - Ir, Pt, Rh, Ru Russia (40%) - Pd	n/a	100%	- Chemical and automotive catalysts - Fuel Cells - Electronic applications
HREEs	P	China (86%) Australia (6%) United States (2%)	China (98%) Other non-EU (1%) UK (1%)	100%	- Permanent magnets for electric motors and electricity generators - Lighting Phosphors - Catalysts
LREEs	P	China (86%) Australia (6%) United States (2%)	China (99%) UK (1%)	100%	- Batteries - Glass and ceramics

Critical Raw Materials

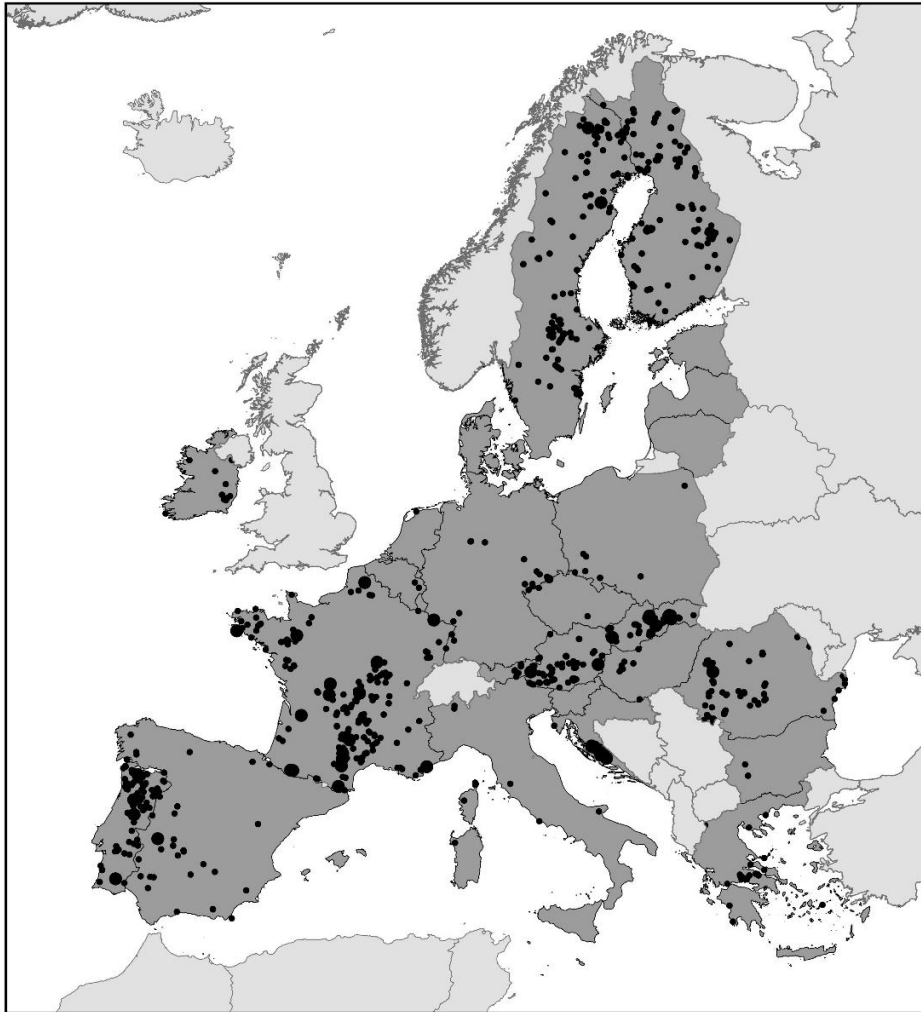


Biggest supplier countries for CRMs



CRM EU-27 deposits

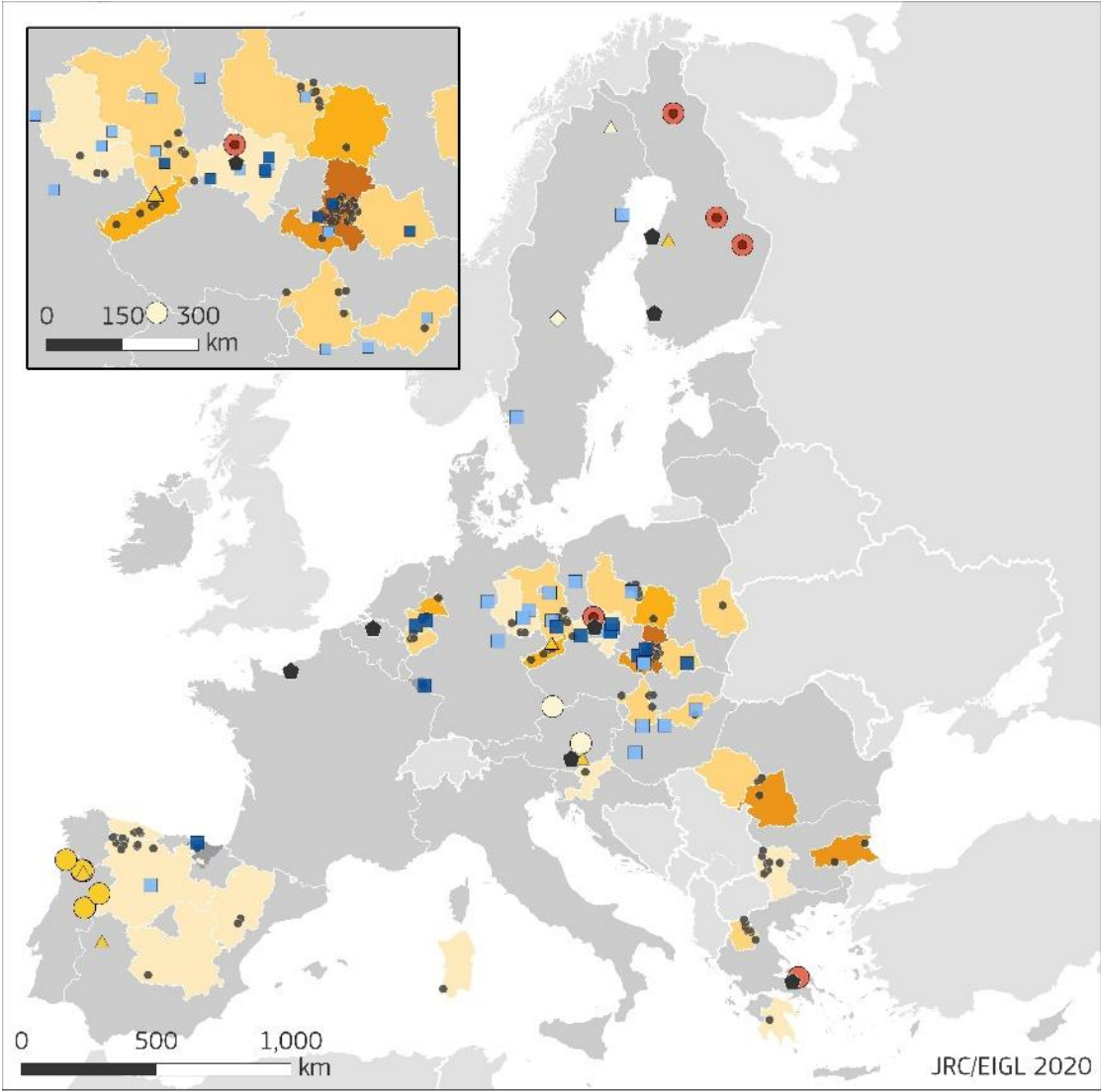
CRITICAL RAW MATERIALS RESOURCES POTENTIAL IN THE EU



Data provided by EuroGeoSurveys combined with other EU data sources

- Europe is well-endowed with aggregates and industrial minerals as well as certain base metals such as copper and zinc
- EU is less successful in developing projects to source critical raw materials, even though there is significant potential for these.
- The reasons are multi-faceted: lack of investment in exploration and mining, diverse and lengthy national permitting procedures or low levels of public acceptance.
- The development of battery raw materials such as Li, Ni, Co, graphite and Mn provides interesting opportunities.

Battery raw material mines, battery factories and coal mines



BATTERY RAW MATERIALS (2017/2018)

- Mines**
- Graphite
 - Lithium
 - Nickel
 - Cobalt (by-product of Ni/Cu)

- Status**
- Production
 - ◇ Preproduction
 - △ Feasibility

- Smelters/refineries**
- ◆ Smelter/refinery

BATTERY FACTORIES (2019)

- Existing (in coal region)
- Future

COAL MINES (2015)

- Operating mine

- Direct jobs in coal mines**
- 80 000
 - 10 001 - 15 000
 - 6 001 - 10 000
 - 1 500 - 6 000
 - ≤ 1 500
 - N.A.

Conflict Minerals

- Conflict minerals are minerals mined in conditions of armed conflict and human rights abuses, and which are sold or traded by armed groups.



Coltan



Cassiterite



Wolframite

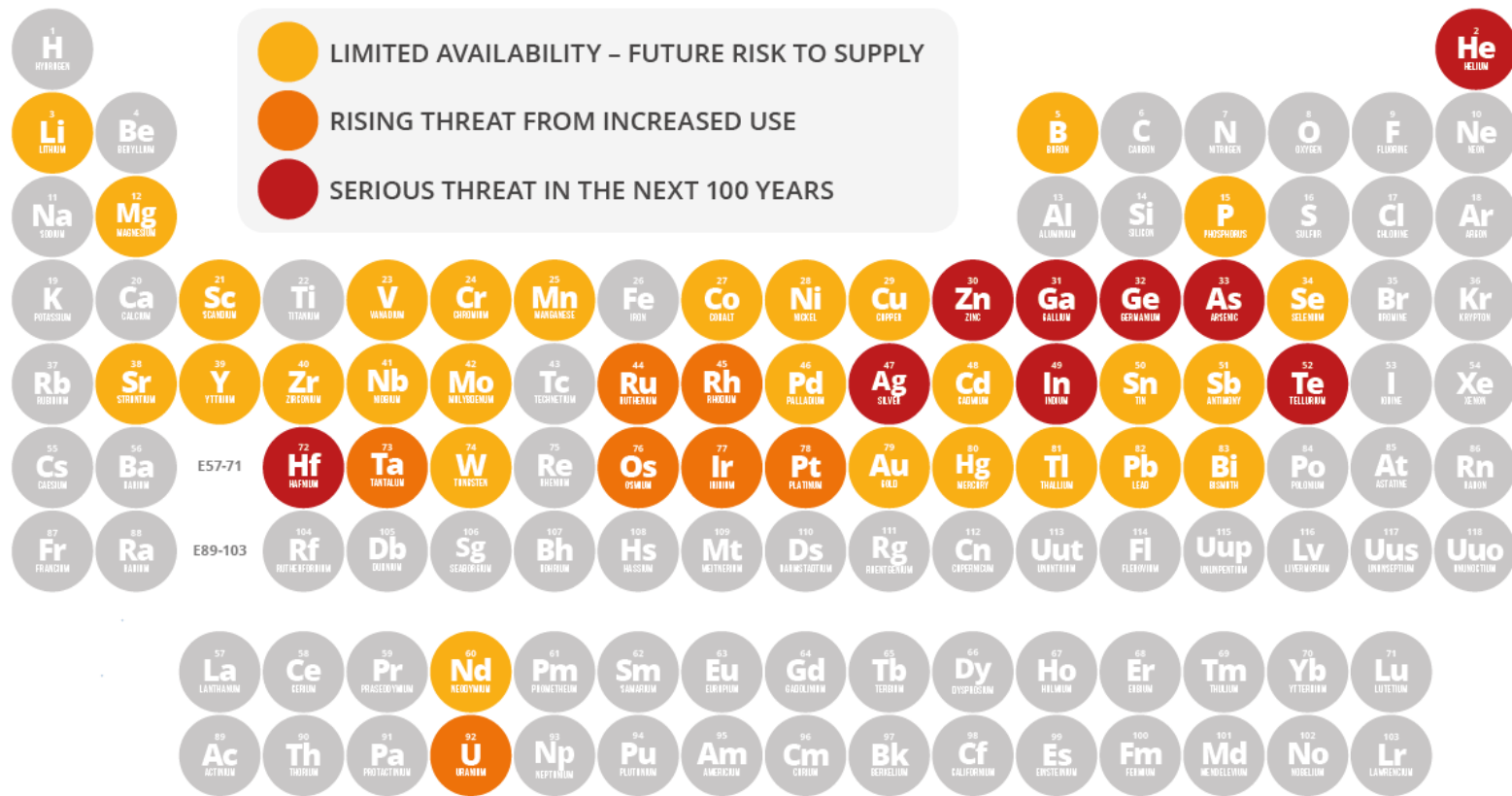


Gold

Conflict Minerals

Mineral	Description	Major uses
Cassiterite	Tin (Sn) is extracted	Plating and solders for joining pipes and electronic circuits
Columbite-tantalite	Tantalum (Ta) is extracted	Electrical components (including those used in mobile phones, computers, videogame consoles), aircraft and surgical components
Wolframite	Tungsten (W) is extracted	Metal wires, electrodes and contacts in lighting, electronic, electrical, heating and welding applications
Gold	Rare metal found in a native (pure) form and obtained as a by-product of other mining operations	Jewellery, electronic, communications and aerospace equipment

THE PERIODIC TABLE'S ENDANGERED ELEMENTS



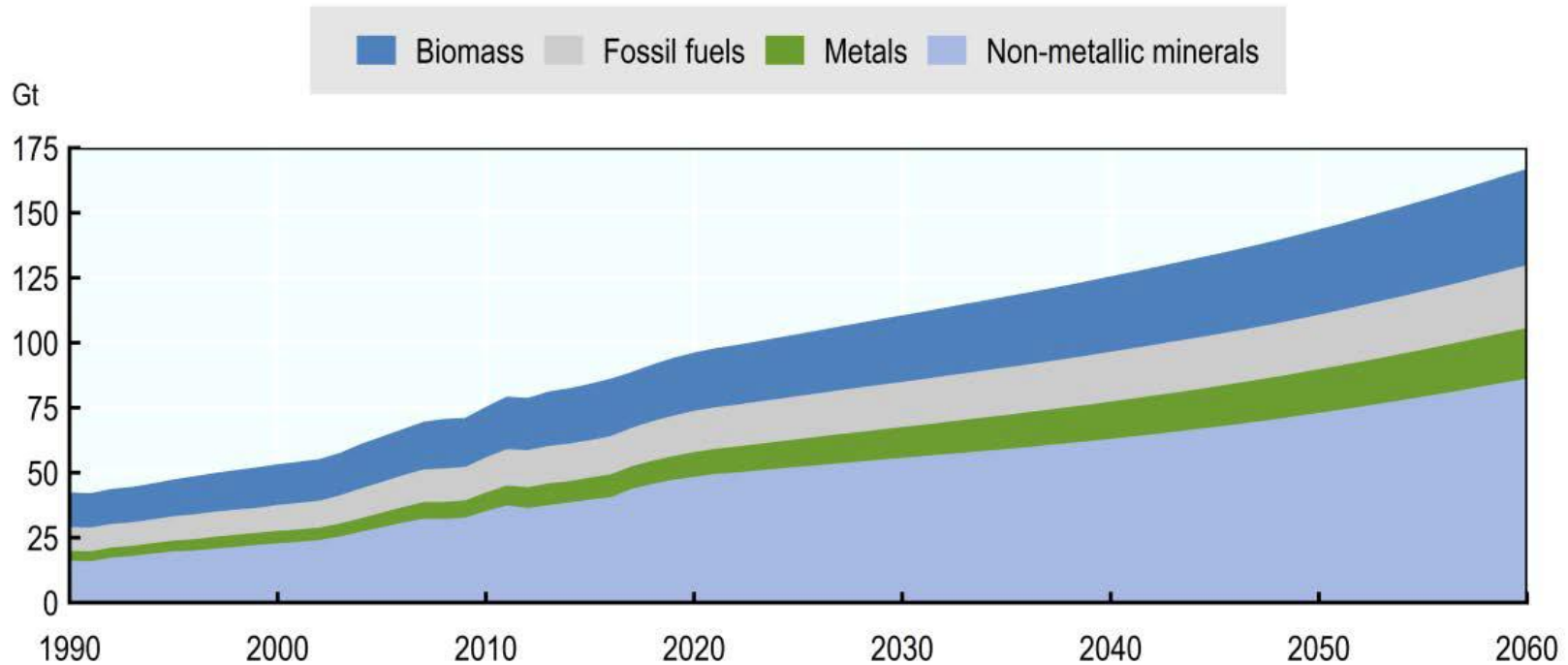
SOURCE: CHEMISTRY INNOVATION KNOWLEDGE TRANSFER NETWORK



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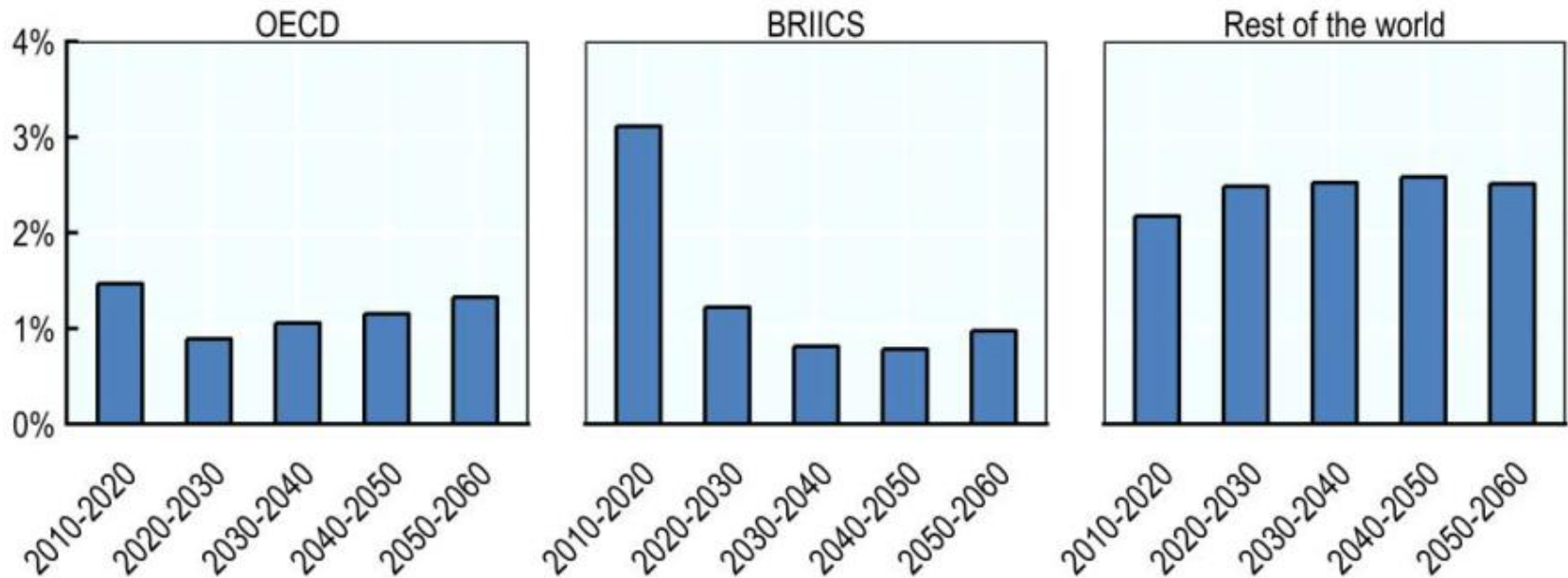
Materials extraction projection by 2060



Global materials extraction is projected to increase across all material types

- Biomass and fossil fuel resources can be directly linked to the coal power and iron and steel production activities; for metals and non-metallic minerals, the projections of materials use are linked to the input of extractive commodities into a processing sector.
- Global materials use is projected to reach 111 Gt in 2030 and 167 Gt in 2060, from 89 Gt in 2017. Thus, total materials extraction is projected to increase by 88% (i.e. almost double) over 43 years.

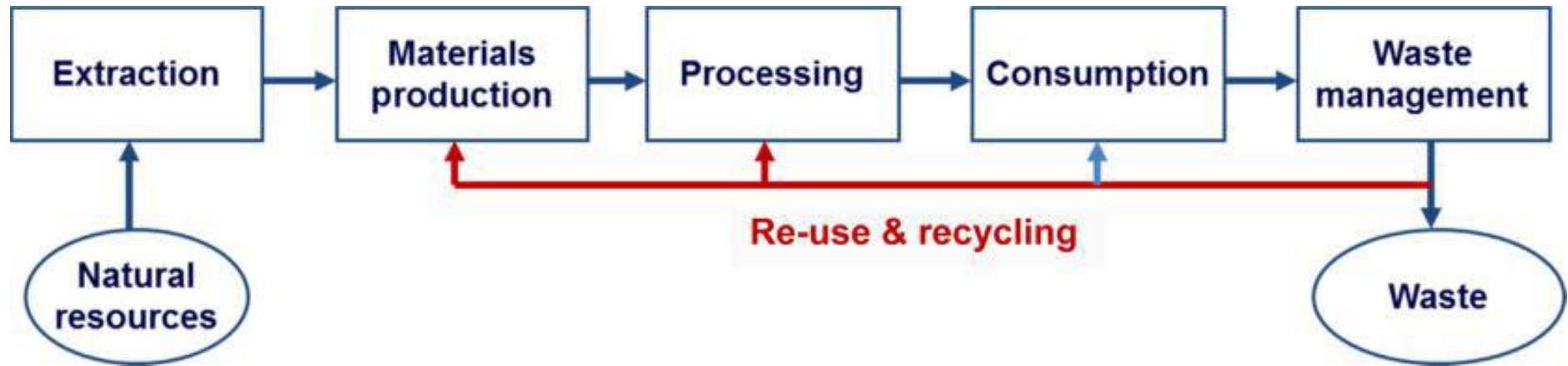
Materials extraction projection by 2060



Average growth rates of materials use by decade for groups of countries

- Growth rates of materials use at the macro-regional level are strongly affected by the economic dynamics in these regions. They are projected to be fairly stable in OECD countries. In the BRIICS countries, growth rates of materials use are projected to be high in the short run, following the growth boom, and then to fall gradually back to levels similar to those of the OECD. The developing countries are projected to continue to catch up and sustain growth rates above those of the OECD throughout the projection horizon.

Recycling: A Key Factor for Resource Efficiency



Closing the loops. Shifting from a linear economy to a circular economy requires the closing of loops to ensure that products and materials are reused or recycled

- Historically, industry has operated as an open system, transforming resources to products that are eventually discarded to the environment. This, has led to growing impacts on the environment.
- Waste is becoming a large problem, as we are running out of land for landfilling, and end-of-life waste treatment has negative environmental and health impacts. This is especially a problem for emerging economies, where material use (and hence discarding of it) is growing very rapidly, while limited waste management infrastructure exists.

Recycling: A Key Factor for Resource Efficiency

- To maintain our level of welfare, services by resources should be provided more efficiently using less (environmental) resources per unit of activity; i.e. we must improve the resource efficiency of our society. There are several ways that we can improve the resource efficiency of society:
 - Use resources more efficiently in the provision of an activity or product (including lifetime lengthening).
 - Use less resource-related services.
 - Reuse product and services.
 - Recycle the resources and materials in products.
- Waste is only waste if it cannot be used again or if its economic value, including dumping costs, is not sufficient to make its exploitation economically feasible. Economic recycling enables waste to become a resource; however, various aspects hinder it becoming totally reusable. Recycling is the reprocessing of recovered materials at the end of product life, returning them into the supply chain.

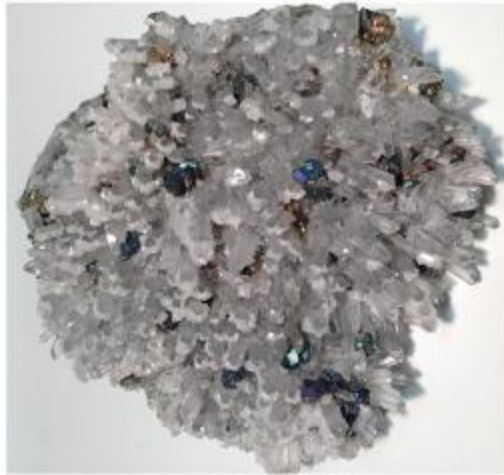
Recycling: A Key Factor for Resource Efficiency

Geological copper minerals

>15 minors e.g. Au, Ag, PGMs, Se

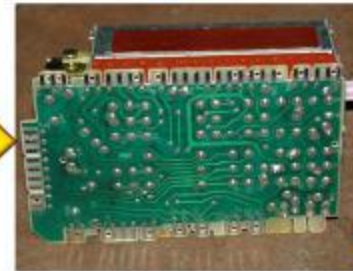
Designed copper “Minerals”

>40 elements complexly linked as alloys, compounds, materials



Geological linkages

Various copper sulphide minerals
on quartz and calcite



Product design & material combinations create new “Minerals”

Functional material connections

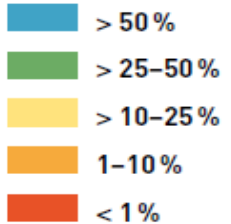
Joined materials multi-material particles

- Conventional extraction processes can recover various elements from geological ores economically, while much work has to be done to recover all metals from complex designer copper “minerals”.
- Less than 1% of the REEs in waste are currently being recycled as these go lost owing to the complexity. The fraction of a material that can reenter the life cycle will depend both on the material itself and on the mineralogy of the product from which it is being recovered.

Functional recycling of metals: ferrous and nonferrous

- Mn:** present at 0.3-1.0% in nearly all steels
- Cr:** chief additional constituent of stainless steel
- Co:** superalloys, catalysts and batteries
- Mo:** high performance stainless steels
- Nb:** high strength-low alloy steels and superalloys
- Zn:** coating steel (galvanizing)
- Cu:** conducting electricity and heat
- Ni:** constituent of stainless steel and superalloys

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* Lanthanides

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
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** Actinides

89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr
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Functional recycling: recycling in which the physical and chemical properties that made material desirable in the first place are retained for subsequent use

Functional recycling of metals: precious metals

Ru: electronics (hard disk drives) and process catalysts /

electrochemistry

Rh, Pd: auto catalysts

Ag: electronics, catalysts, batteries, glass mirrors

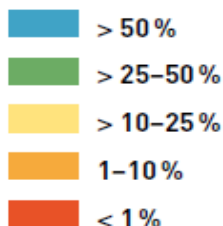
Ir: electrochemistry, crucibles for monocrystal growing

Os: catalysts, has little industrial importance

Pt: catalysts

Au: electronics

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Functional recycling: recycling in which the physical and chemical properties that made material desirable in the first place are retained for subsequent use

Functional recycling of metals: specialty metals

Zr, Hf: nuclear reactors

In: coating in flat panel displays

Cd: batteries (85%)

Ga: electronics – LED, diodes, solar cells

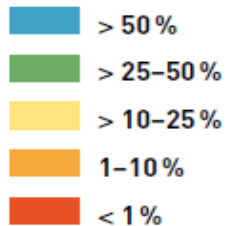
Ge: night vision (IR) lenses, PET catalysts, fiber optics, solar cells concentrators

Sr: pyrotechnics, ferrite ceramic magnets for electronics

Se: glass manufacture, LEDs, photovoltaics, IR optics

As: semiconductors (electronics, PVs), alloying element

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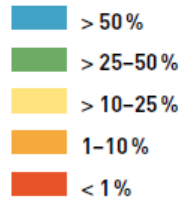
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Specialty metals: present in small amounts for their specific physical and chemical properties

End of Life-recycling rate for metals

EOL-RR: reflects the total material input into the production system that comes from recycling of post-consumer scrap

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
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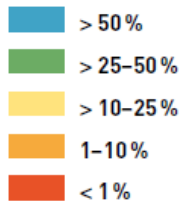
** Actinides

- It is noteworthy that for only 18 of the 60 metals are the EOL-RR values above 50%. Another three metals are in the 25-50% group, and three more in the 10-25% group. For a very large number, little or no EOL recycling is occurring, either because it is not economic, or no suitable technology exists

Recycling input ratio for metals

RIR: the input of secondary material from old scrap to the total input of material (primary and secondary)

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
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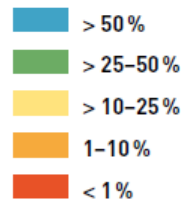
** Actinides

- Pb, Ru and Nb are the only metals for which RIR >50%, but 16 metals have RIR in the >25-50% range. This reflects a combination in several cases of efficient employment of new scrap as well as better than average EOL recycling.

Old scrap ratio for metals

OSR: quantify the share of old (or end-of-life, EoL) scrap in the overall recycling flow

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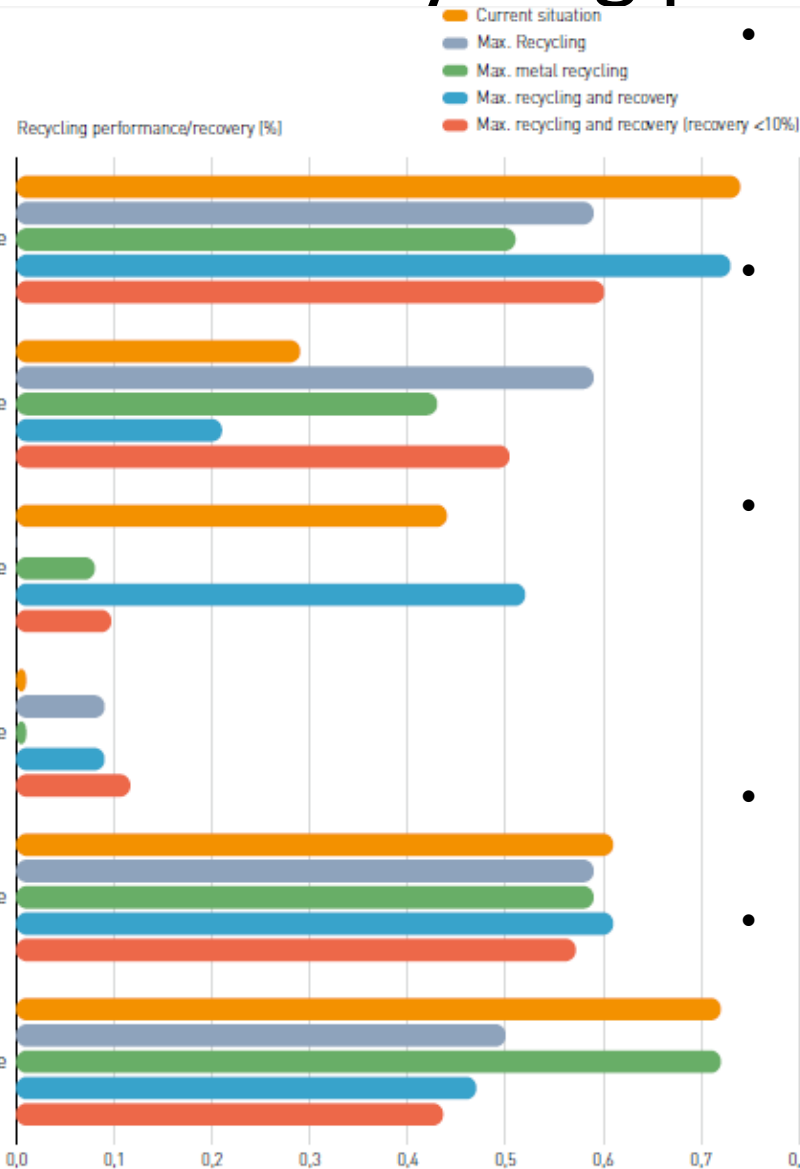
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- OSR results tend to be high for valuable materials, because they are utilized with minimal losses in manufacturing processes and collected at EOL with relatively high efficiency.
- Collection and recycling at EOL are high for the hazardous metals Cd, Hg, Pb. Overall, 13 metals have OSR >50%, and another 10 have OSR >25-50%.

Recycling performance



- EoL recycling rates can be vastly different, depending on the economic actuators used, and whether they focus on material or energy recovery.
- Economics rather than simple legislation-imposed rates should drive recycling, as legislation cannot capture fully the complexity of a recycling system.
- There is more than one recycling rate for metals in a product. Depending on whether economics, processing technology, etc., are the main motivation for recycling, the recycling rates will be different.
- These rates reflect a statistical-distribution range.
- One-dimensional linear recycling-rate definitions are limiting cases, as they cannot consider the complex non-linear interactions found in most recycling systems.

Compatibility matrix as a function of metallurgical recovery



- Depending on the process route followed, high recovery rates as well as high losses are possible.
- A careful attention to design, infrastructure, legislation etc. is necessary.
- This is especially possible for metals closely linked where one metal can be recovered while the other due to this selection of recovery then goes lost. This is driven by thermodynamics, technology, design etc.

Compatibility matrix as a function of metallurgical recovery

Who gets the trash?



- WEEE constitutes about 8 % of municipal waste
- WEEE represents the fastest-growing waste stream in the EU, generating about 8.7 million tonnes in 2005. By 2020, this amount is estimated to reach 12.3 million tonnes, corresponding to an annual growth of about 2.6 %
- The e-waste stream also contains substantial amounts of metals, such as rare earths, lithium (batteries), ruthenium, antimony and tin. Most of the production of indium is used in electronic appliances.
- In many cases, sometimes despite legislation, small WEEE articles are not collected separately for recycling, but disposed of with mixed Municipal Solid Waste

ELEMENTS OF A SMARTPHONE

ELEMENTS COLOUR KEY: ● ALKALI METAL ● ALKALINE EARTH METAL ● TRANSITION METAL ● GROUP 13 ● GROUP 14 ● GROUP 15 ● GROUP 16 ● HALOGEN ● LANTHANIDE

SCREEN



Indium tin oxide is a mixture of indium oxide and tin oxide, used in a transparent film in the screen that conducts electricity. This allows the screen to function as a touch screen.



The glass used on the majority of smartphones is an aluminosilicate glass, composed of a mix of alumina (Al_2O_3) and silica (SiO_2). This glass also contains potassium ions, which help to strengthen it.



A variety of Rare Earth Element compounds are used in small quantities to produce the colours in the smartphone's screen. Some compounds are also used to reduce UV light penetration into the phone.

ELECTRONICS



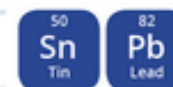
Copper is used for wiring in the phone, whilst copper, gold and silver are the major metals from which microelectrical components are fashioned. Tantalum is the major component of micro-capacitors.



Nickel is used in the microphone as well as for other electrical connections. Alloys including the elements praseodymium, gadolinium and neodymium are used in the magnets in the speaker and microphone. Neodymium, terbium and dysprosium are used in the vibration unit.



Pure silicon is used to manufacture the chip in the phone. It is oxidised to produce non-conducting regions, then other elements are added in order to allow the chip to conduct electricity.



Tin & lead are used to solder electronics in the phone. Newer lead-free solders use a mix of tin, copper and silver.

BATTERY



The majority of phones use lithium ion batteries, which are composed of lithium cobalt oxide as a positive electrode and graphite (carbon) as the negative electrode. Some batteries use other metals, such as manganese, in place of cobalt. The battery's casing is made of aluminium.

CASING



Magnesium compounds are alloyed to make some phone cases, whilst many are made of plastics. Plastics will also include flame retardant compounds, some of which contain bromine, whilst nickel can be included to reduce electromagnetic interference.

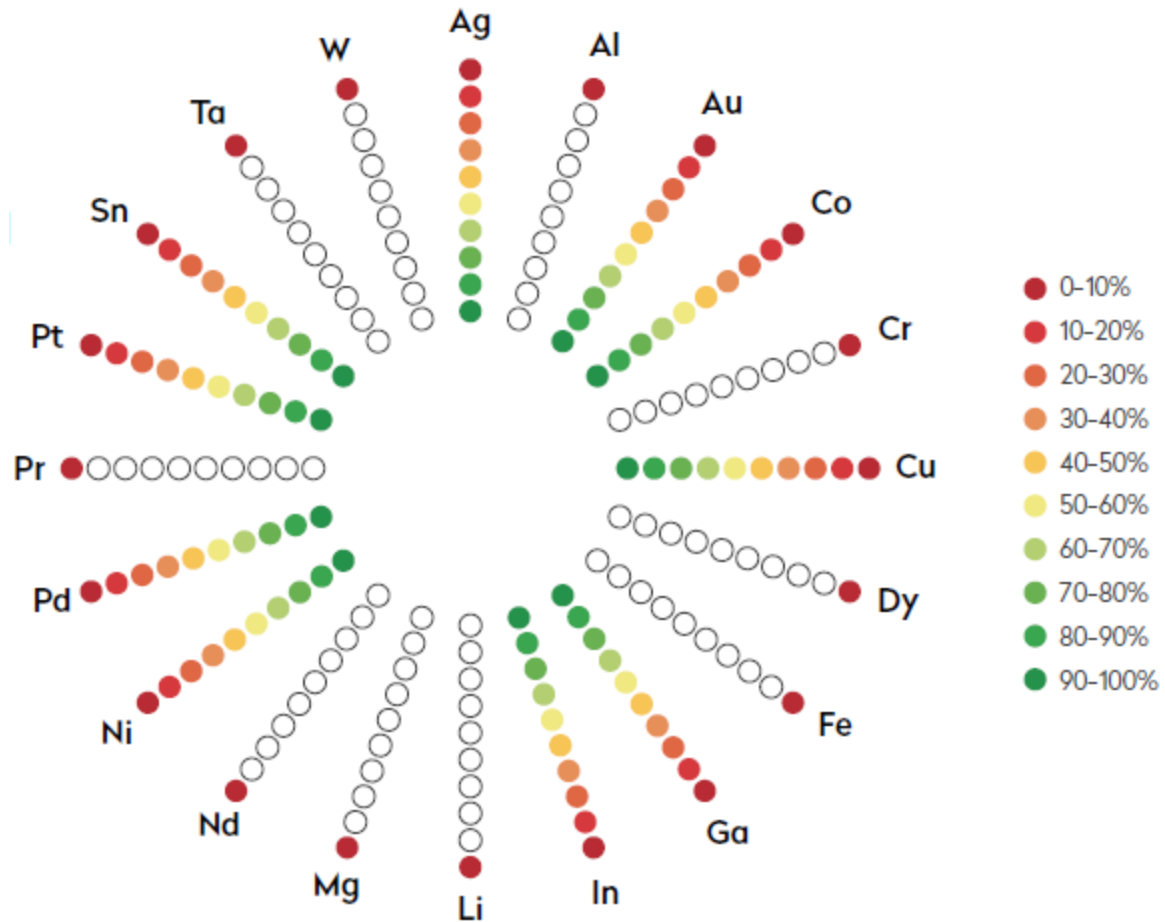
Mobile phone - recycle

- Recycling electronic devices requires a combination of steps, including dismantling, size reduction (shredding), physical sorting and further metallurgical and other final treatment processing
- Materials are inevitably lost at every phase of the process, and the overall recycling efficiency also depends on the design of the products, the properties of the materials they are made of, how well the waste is collected and sorted
- Recycling consumes energy, meaning recovered materials still come at a cost to the environment
- For better recycling, one has to keep the complexity of intermediate materials to its lowest possible value. Creating complex mixtures of scrap and recycle products negatively affects final recovery

Mobile phone recycle - smelting

- Phone is introduced into a high temperature metallurgical reactor and materials are recovered mainly as metals, alloys and inorganic compounds
- This route enables the recovery of all metals that have a high affinity for copper (e.g. gold, silver, palladium and copper itself) while less noble metals such as magnesium, aluminum, lithium and tungsten and its alloy elements end in slag. All plastic reports to offgas as CO₂, H₂O, NO_x etc.
- The advantage of this approach is that the energy from the burning of the plastics saves the use of primary fuels.

Mobile phone recycle - smelting

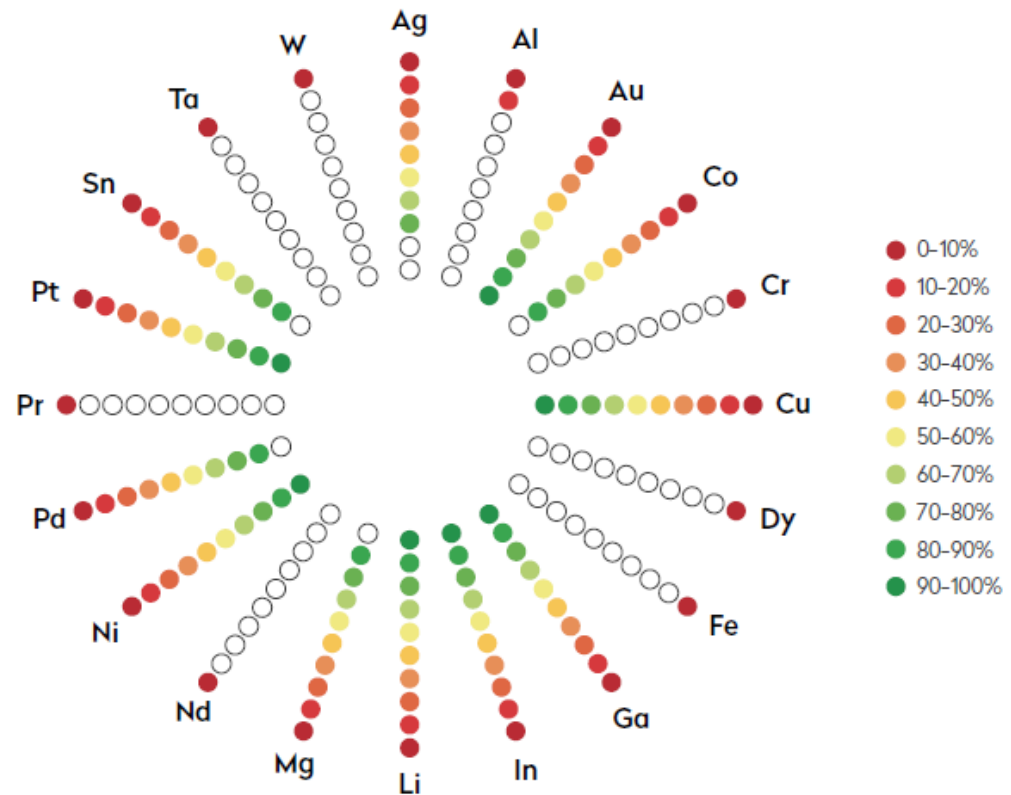


Mobile phone recycle – dismantling and selective smelting

- The second route looked at selective smelting, which includes separating the phone modules and putting them through the most suitable metallurgical and plastic recovery processes
 - Camera, core, top and bottom modules go to top submerged lance (TSL) furnace
 - Display module goes to light metal remelt/refine, for optimal recovery of materials such as magnesium
 - Battery goes to an electric furnace, for optimal recovery of materials like lithium and cobalt
 - Back cover goes to the plastic extruder, for optimal recycling of the polycarbonate (plastic)
- This route tests the effect of modularity

Mobile phone recycle – dismantling and selective smelting

- Route 2 delivers the widest range of recovered materials, and in contrast to route 1, also successfully recovers plastics in a usable form because the plastic case is sent directly to the plastics extruder. It also results in successful recovery of magnesium (90%), which is completely lost in the first route

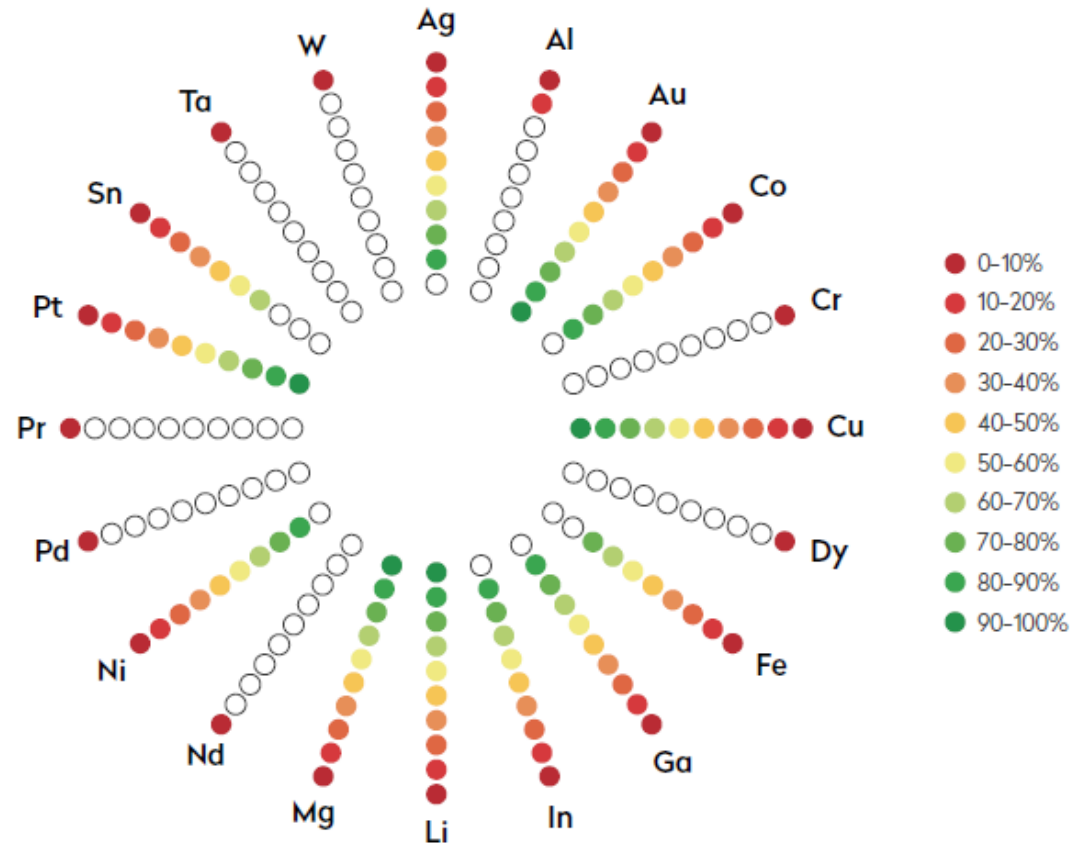


Mobile phone recycle – Shredding, physical pre-processing and metallurgy

- The final route involves removing the battery and feeding the remainder through a cutting mill, followed by separating the small pieces (scrap) into different valuable scrap fractions.
- The separation is achieved by using the physical properties of the elements, for example using the different degrees of conductivity and magnetism to separate iron and other ferrous metals from the non-ferrous non-magnetic materials (e.g. copper, aluminum and magnesium). Optical sorting is also used to separate plastic pieces from metal-rich pieces
- These fractions are processed using the appropriate technology either metallurgically, for energy recovery (mostly from burning polymers), or into plastic products

Mobile phone recycle – Shredding, physical pre-processing and metallurgy

- Route 3 is the best for recovering the most non-precious metals (e.g. Al, Mg) and results in the highest overall recovery rates by weight.
- It is less efficient in recovering valuable metals like Pd or Sn as these go lost for example in complex plastic and bulk metal scrap mixtures especially when compared to route 2. It also produces impure mixed plastic recyclate which cannot be easily reused for consumer applications



Mobile phone recycle

- Which of the currently available recycling processes (including physical separation, metallurgical, energy recovery, plastics processing) could offer the best material recovery rates for recycling the mobile phone?
- How modularity could contribute to improving recyclability?
- Look at three metrics, (1) metal recycling rate, (2) total material recycling rate (which includes the previous) and (3) total recycling and recovery rate (materials and energy).

Mobile phone recycle

- Route 1 (smelting) offered the lowest percentage of recovered materials by weight
 - 14% metal recycling,
 - 25% total material recycling
 - 36% recovery (= recycling + energy recovery) for all materials,
 - a poor range of materials recovered.
- Route 2 (dismantling/selective smelting) offered greater recovery of materials by weight
 - 19% metal recycling,
 - 28% total material recycling
 - 31% recycling/recovery
 - the widest variety of materials recovered.

Mobile phone recycle

- Route 3 (shredding) offered the highest percentage of materials by weight
 - 22% metal recycling
 - 30% total material recycling
 - 31% recycling and recovery
 - the variety of materials recovered is more limited than route 2 due to the creation of complex mixtures, dust etc. by shredding/cutting.
- What are the optimum steps for processing of a product?
- The more separation steps, the more residues. When recycling a mono-material product, the steps are simple; for a mobile phone you need many more steps to recover as many of the 46 elements as possible.

Mobile phone recycle

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Mobile phone recycle

- When selecting the preferred recycling route, it is insufficient to look at total weight alone, in part because the higher weight (in route 3, for example) is contributed primarily to non-precious materials.
- We also need to consider the variety of materials recovered because of the scarcity and/or potential sourcing issues of high-value (precious) metals that are present in the phone in much smaller weights.
- Based on both total weight and variety of materials recovered, recycling route 2 appears to be the best option for recycling

Benefits from increased recycling



Co-funded by the
Erasmus+ Programme
of the European Union



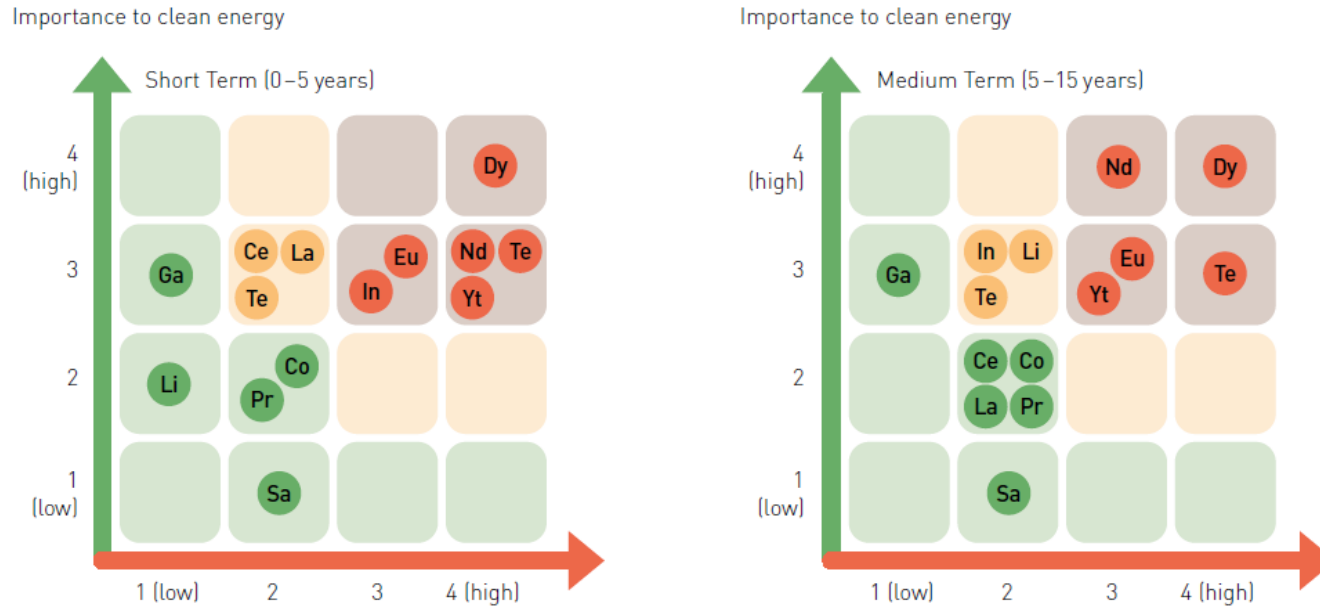
Value of recovered metals

- Metal prices directly or indirectly influence the financial rewards of recovery.
- These are related to:
 - the physics of primary and recycled recovery of a metal
 - the relative abundance of the various elements in primary minerals
 - the demand for greater sustainability and other services provided by the metal
- The precious metal content of various devices can do much to boost the economic recycling of WEEE products
 - the metal value of used rechargeable batteries is significant, driven mainly by Co and Ni, which in 2010 had average prices of \$45 and \$21 per kg (Ni contained in one tonne of NiMH batteries has a value of about \$6,000)
 - more value can be extracted from recycled batteries, if their materials are recovered as compounds instead of breaking them down to an elemental state. However, this obviously requires a market for these compounds to flow back into the products, implying that these could have a high market value and thus rendering the recycling economic

Value of energy saved in metal production

- Improved recycling processes can be much cheaper than primary production, because they can use much less energy in production of the metals.
 - significant lowering in energy consumption required for the recycling of steel through Electric Arc Furnace smelting (9 to 12.5 GJ/tonne) compared to:
 - primary production from a Blast Furnace- Open Hearth Furnace (26.4 to 41.6 GJ/tonne)
 - Blast Furnace-Basic Oxygen Furnace (19.8 to 31.2 GJ/tonne)
 - Direct Reduction-Electric Arc Furnace (28.3 to 30.9 GJ/tonne).
 - A more sophisticated approach to recycling can help reducing the energy use of existing recycling processes, for example for steel. Scrap with impurities increases the energy required for producing metal of a required purity. Integrating new technology into existing processes may deal with some of these impurities; a metal-recycling system that capitalizes on this can maximize profits.

Reducing the risk of metal scarcity, particularly of metals needed for transition to a green economy



- Future supply may be compromised for CRMs due to resource depletion, poor design, a 'throw-away' society, cheap landfill costs, insufficient policy, insecurity of supply through geopolitical risks, insufficient development of current and future technologies, etc.
- Many of CRMs enable sustainability in future products and infrastructure required by a low-carbon, resource-efficient, sustainable society, including energy production (e. g. solar, wind, smart grids), water purification (e. g. enhancing water quality by using sensors, filter materials, smart water systems), transportation (e. g. electric cars, planes), and construction (e. g. various materials in eco-cities).